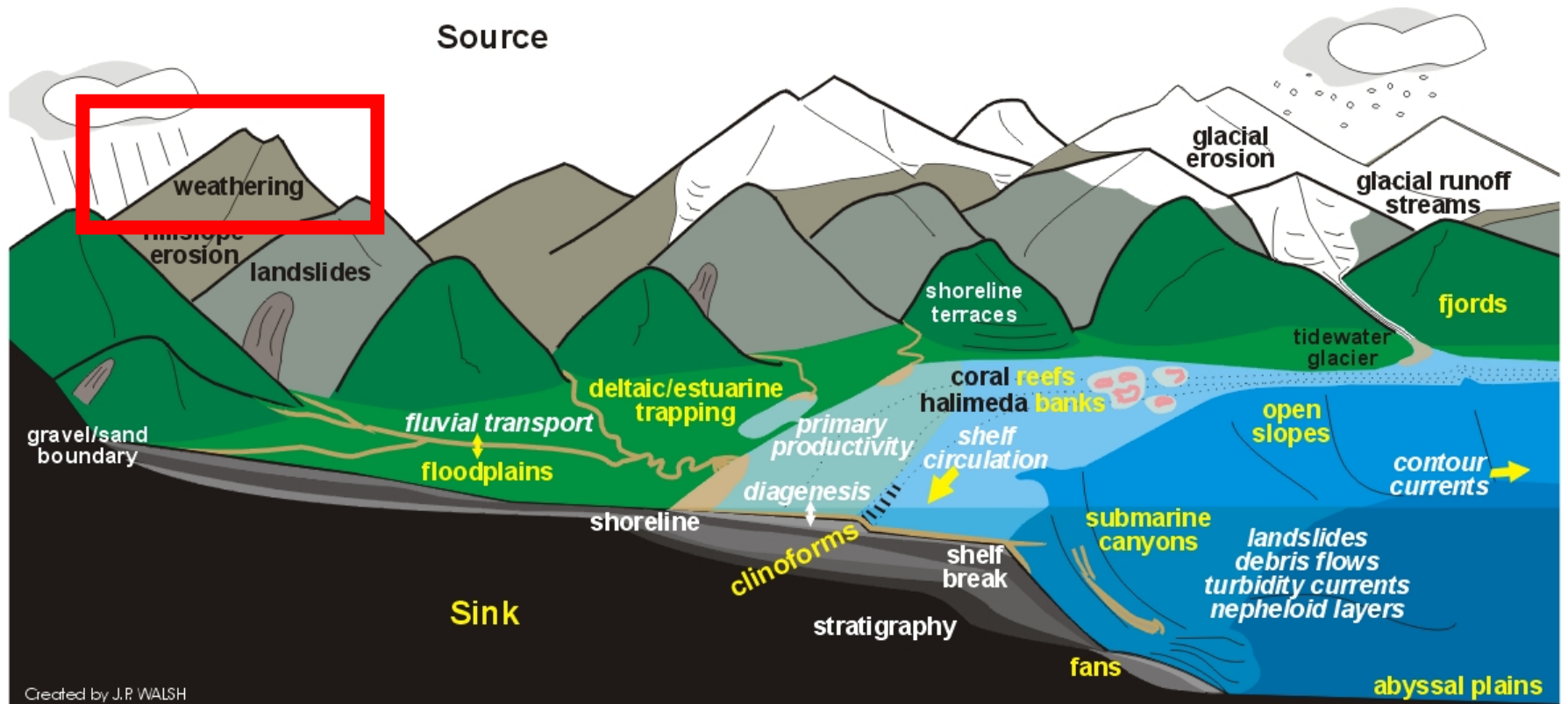
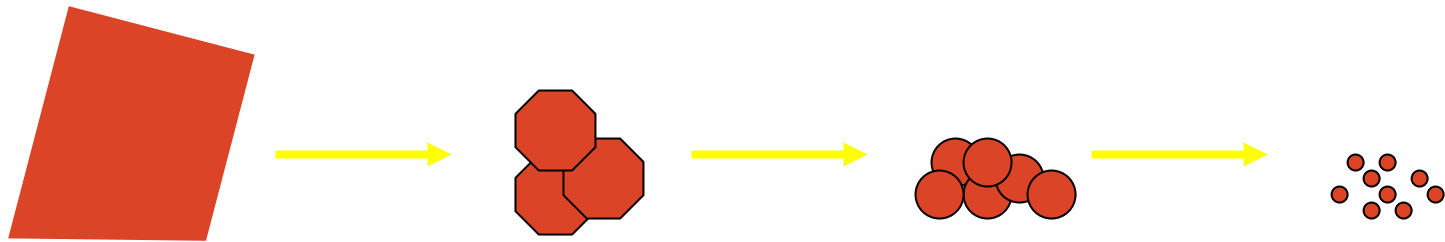


Weathering

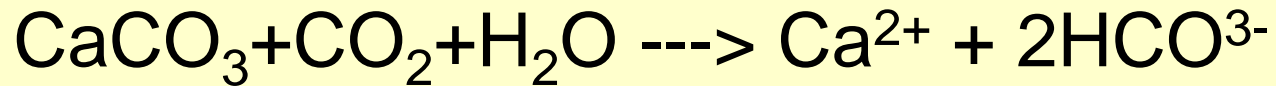


Weathering: the disintegration,
or breakdown of rock material

Mechanical Weathering: no change in chemical composition--just disintegration into smaller pieces



Chemical Weathering: breakdown as a result of chemical reactions



Mechanical Weathering

Physical breakup

- pressure release
- water: freeze - thaw cycles
- crystallization of salt in cracks
- thermal expansion and contraction

All this increases the total surface area exposed to weathering processes.

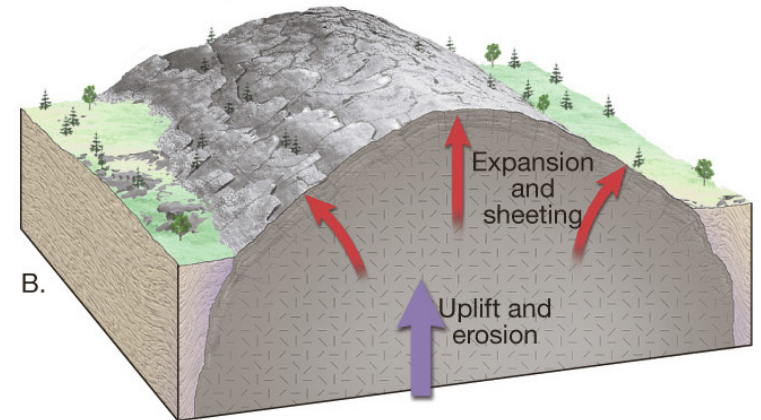
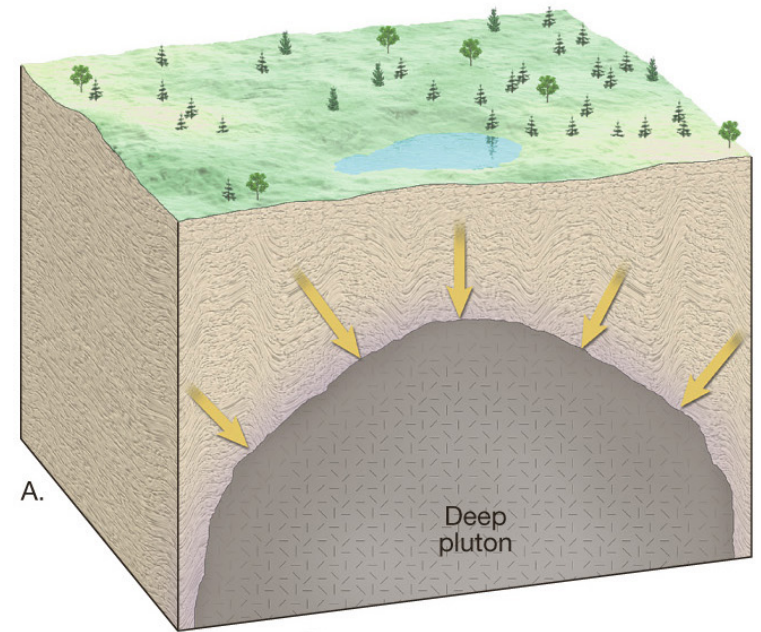
Mechanical Weathering

Exfoliation:

Rock breaks apart in layers that are parallel to the earth's surface; as rock is uncovered, it expands (due to the lower confining pressure) resulting in exfoliation.



Mechanical Weathering



Sheet Joints (Exfoliation)





Half Dome,
Yosemite, CA

GA

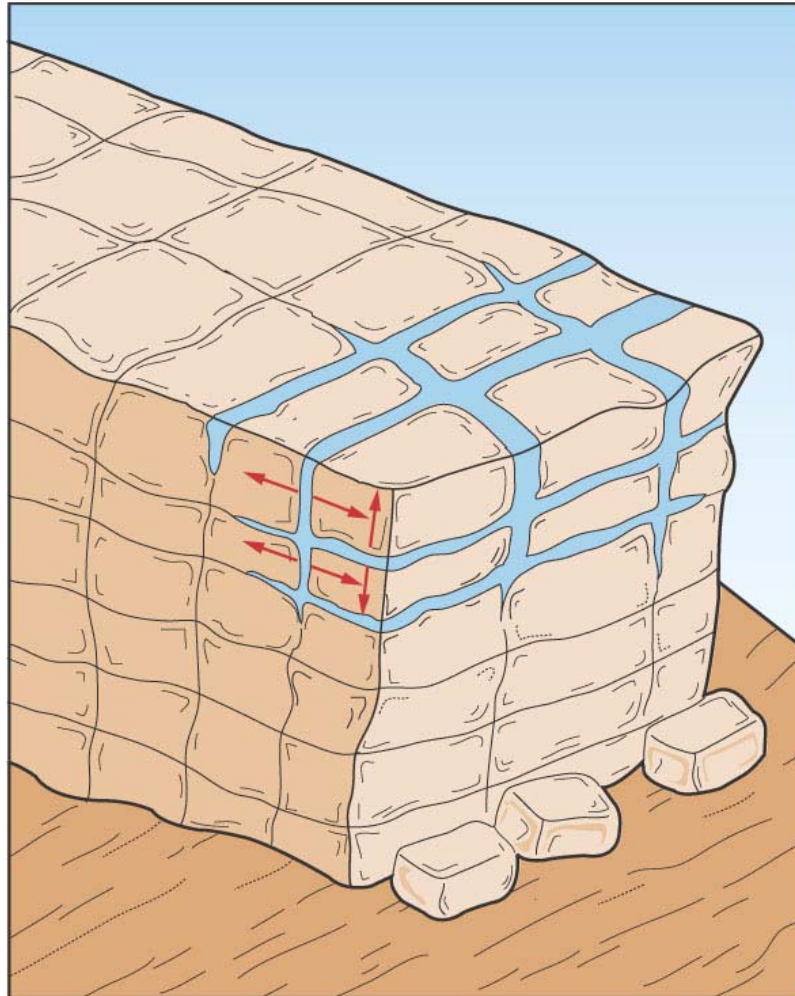
Exfoliated Domes, Yosemite



Stone Mountain, Georgia, showing the product of exfoliation due to unloading



Frost Wedging: rock breakdown caused by expansion of ice in cracks and joints



Shattered rocks are common in cold and alpine environments where repeated freeze-thaw cycles gradually pry rocks apart.



Thermal expansion due to the extreme range of temperatures can shatter rocks in desert environments.

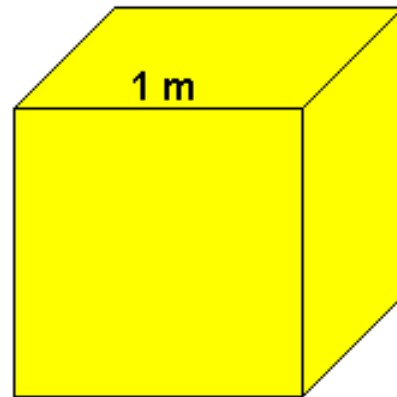
Repeated swelling and shrinking of minerals with different expansion rates will also shatter rocks.



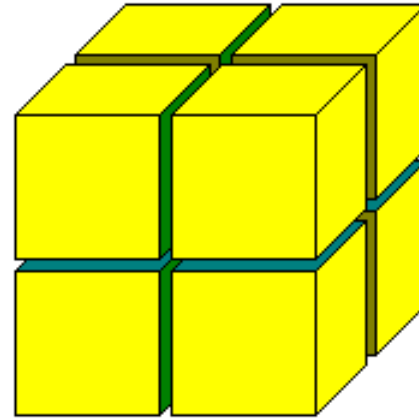
Role of Physical Weathering

- 1) Reduces rock material to smaller fragments that are easier to transport
- 2) Increases the exposed surface area of rock, making it more vulnerable to further physical and chemical weathering

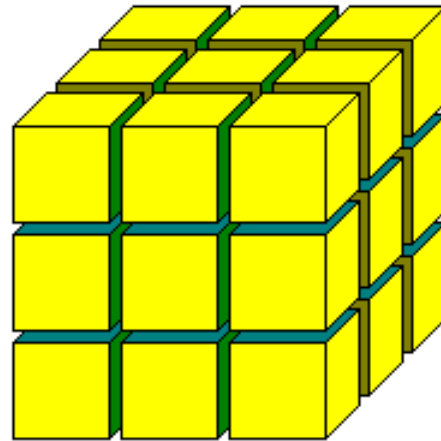
Surface Area and Weathering



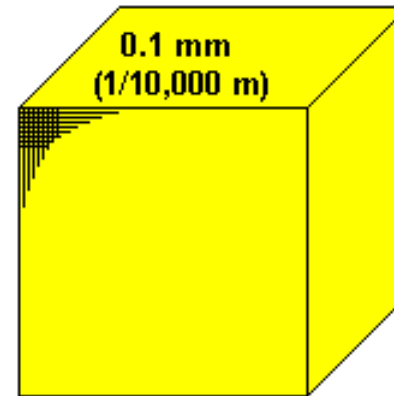
$$\text{Area} = 6 \times 1\text{m}^2 = 6 \text{ m}^2$$



$$\text{Area} = 6 \times (1/2\text{m})^2 \times 8 = 12 \text{ m}^2$$



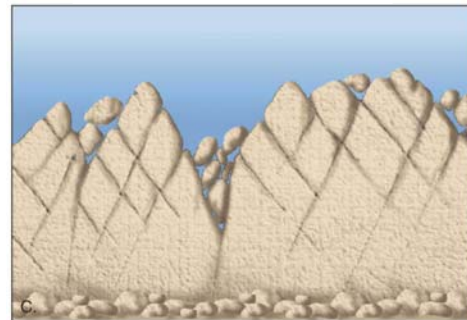
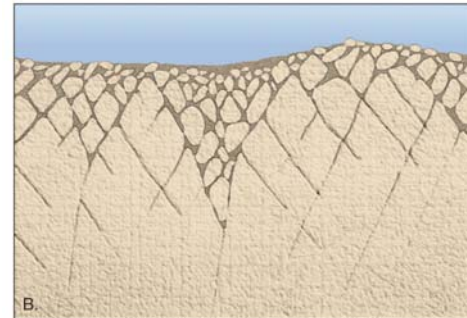
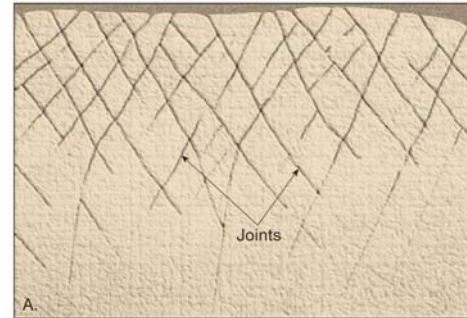
$$\text{Area} = 6 \times (1/3\text{m})^2 \times 27 = 18 \text{ m}^2$$



$$\text{Area} = 6 \times (1/10,000\text{m})^2 \times 10^8 = 10^4 \text{ m}^2 = 2.5 \text{ acres}$$

Rates of weathering

Joints in a rock are a pathway for water - they can enhance mechanical weathering



D.

Chemical Weathering

Definition: transformation/decomposition of one mineral into another

Mineral breakdown

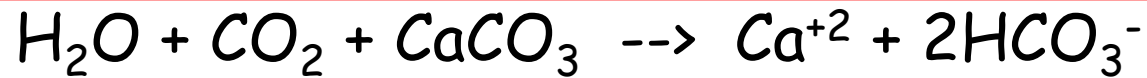
- carbonate dissolves
- primary minerals --> secondary minerals (mostly clays)

Net loss of elements retained in the soil.

Chemical Weathering

- Water is the main operator:
 - Dissolution
 - Many ionic and organic compounds dissolve in water
 - Silica, K, Na, Mg, Ca, Cl, CO_3 , SO_4
 - Acid Reactions
 - Water + carbon dioxide \leftrightarrow carbonic acid
 - Water + sulfur \leftrightarrow sulfuric acid
 - H^+ effective at breaking down minerals

Dissolution



water + carbon dioxide + calcite
dissolve into calcium ion
and bicarbonate ion

Biological activity in soils
generates substantial CO_2

Bicarbonate is the dominant
ion in surface runoff.



Chemical Weathering

- Oxidation
 - Oxygen dissolved in water promotes oxidation of sulfides, ferrous oxides, native metals
- Organic Activity
 - Plant material makes H^+ ions available

Chemical Weathering

- Hydration: attachment of water molecules to crystalline structure of a rock, causing expansion and weakness
- Hydrolysis: combination of hydrogen and oxygen in water with rock to form new substances

Chemical Weathering

Solution: process by which rock is dissolved in water

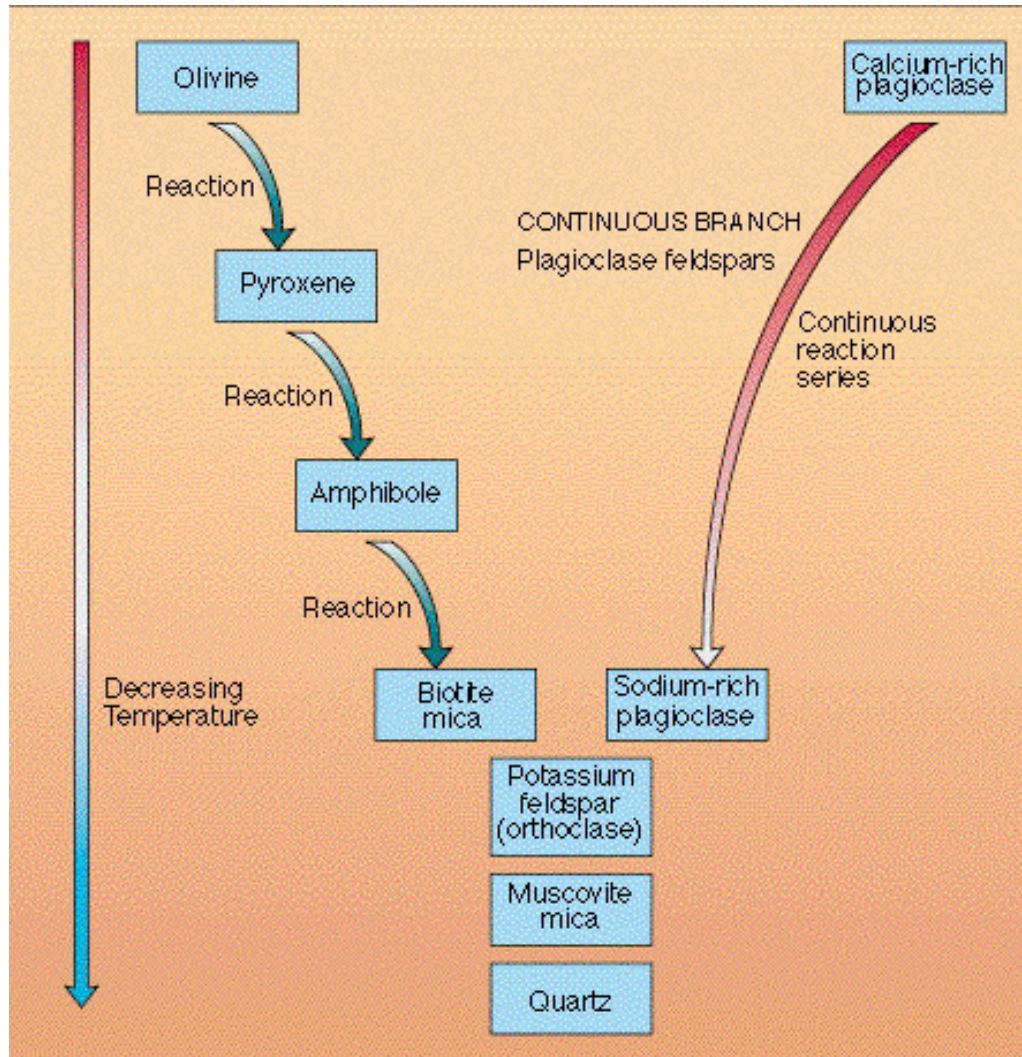
- Is strongly influenced by pH and temperature
- When water becomes saturated, chemicals may precipitate out forming *evaporite* deposits.
- Calcium carbonate (calcite, limestone), sodium chloride (salt), and calcium sulfate (gypsum) are particularly vulnerable to solution weathering.

Resistance to Weathering

First to Crystallize

Bowen's Reaction Series

Last to Crystallize

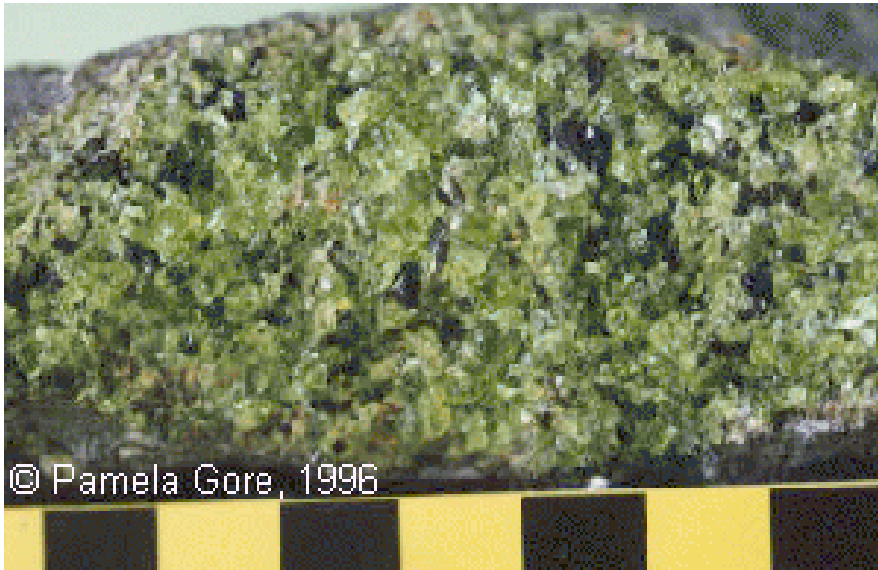


Fast Weathering

Goldrich Stability Series

Slow Weathering

Olivine/pyroxene to clay



+ H_2CO_3 (acid)



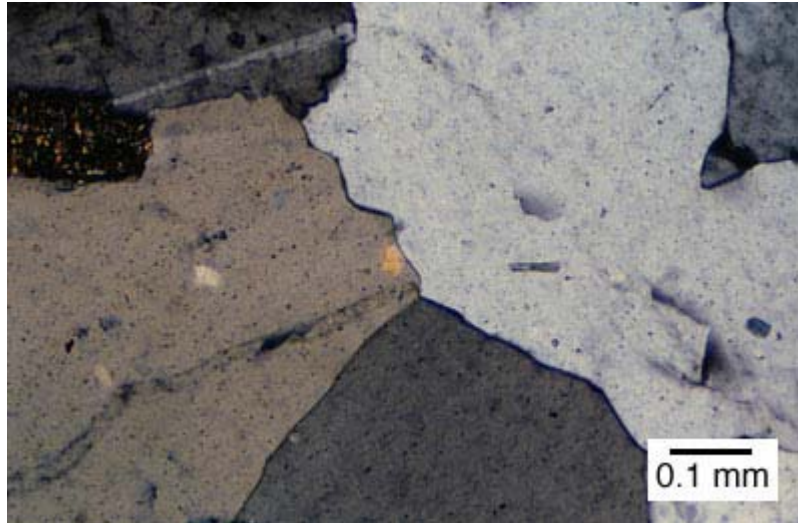
Feldspars to clay



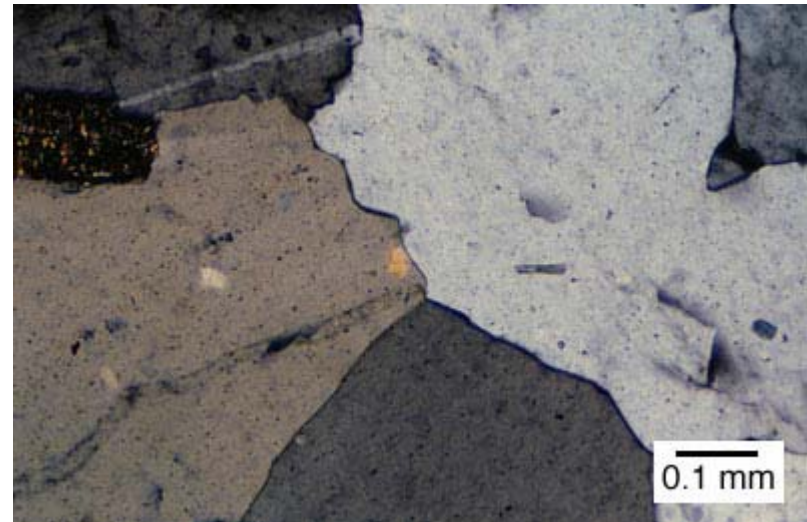
+ H_2CO_3 (acid)



Quartz to quartz (!)



+ anything



Calcite to

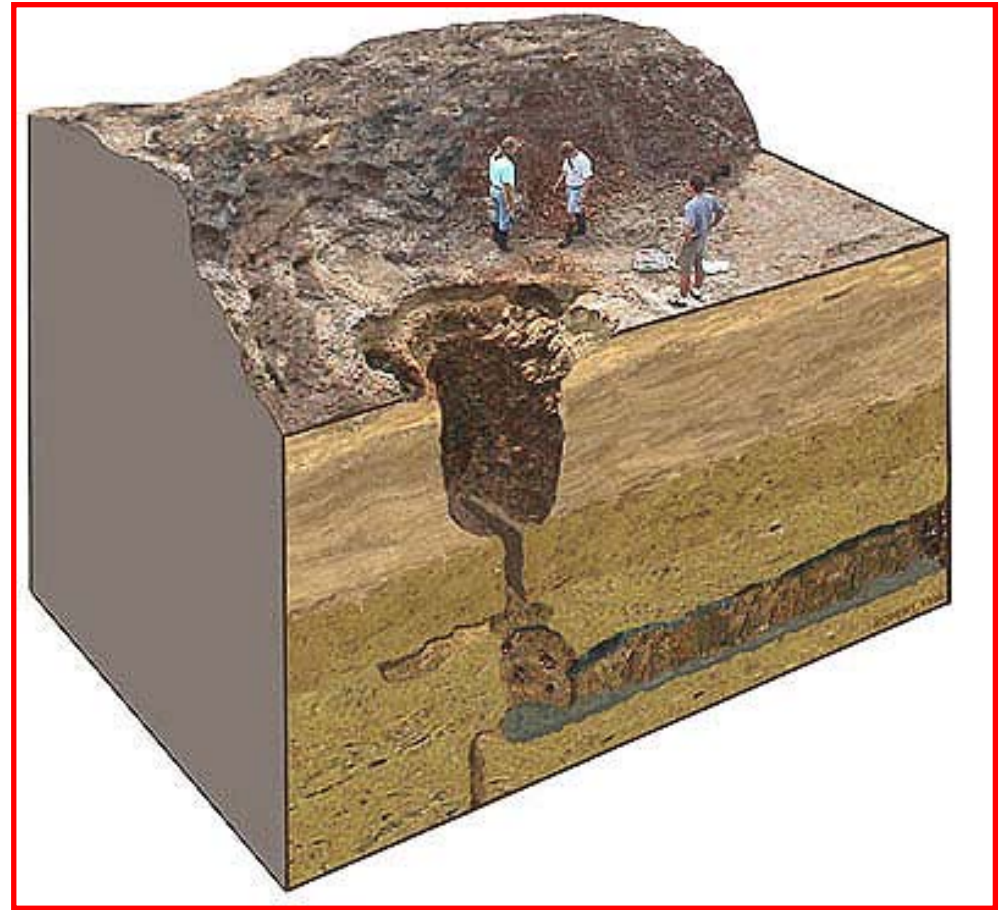


+ anything



nothing

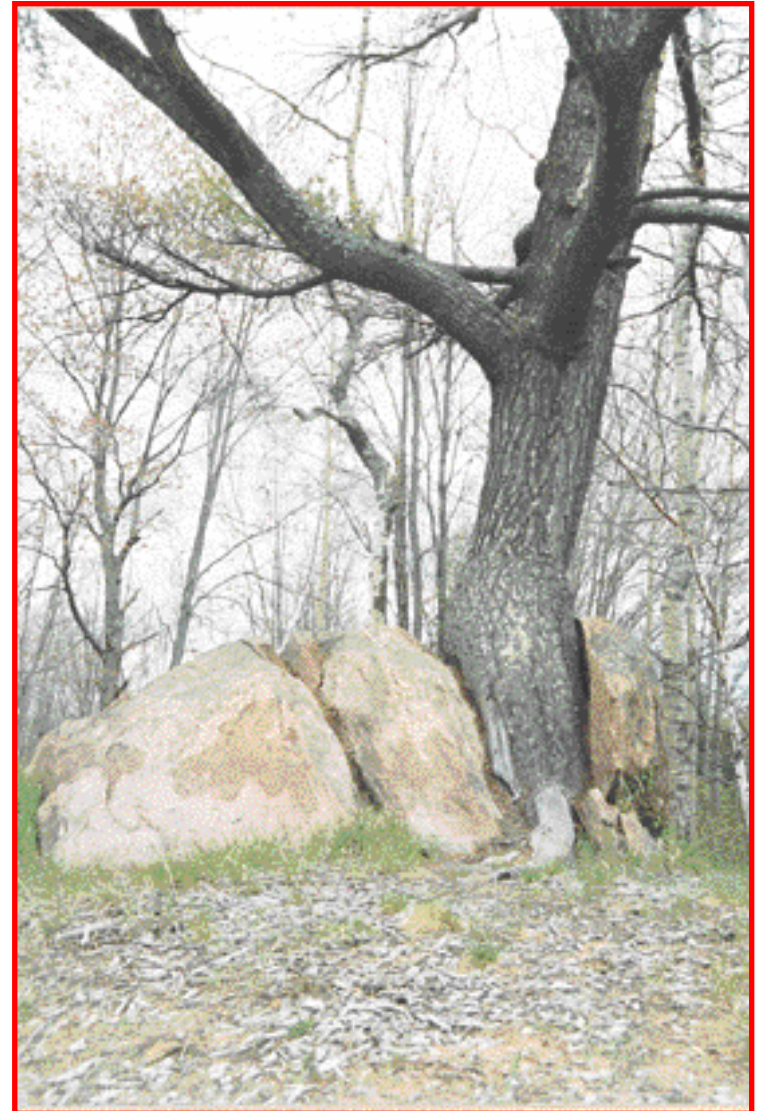
'Karst' landforms develop in areas underlain with limestone



Biological Weathering

Can be both chemical and mechanical in nature.

- roots split rocks apart
- roots produce acids that dissolve rocks.
- tree throw
- burrowing animals



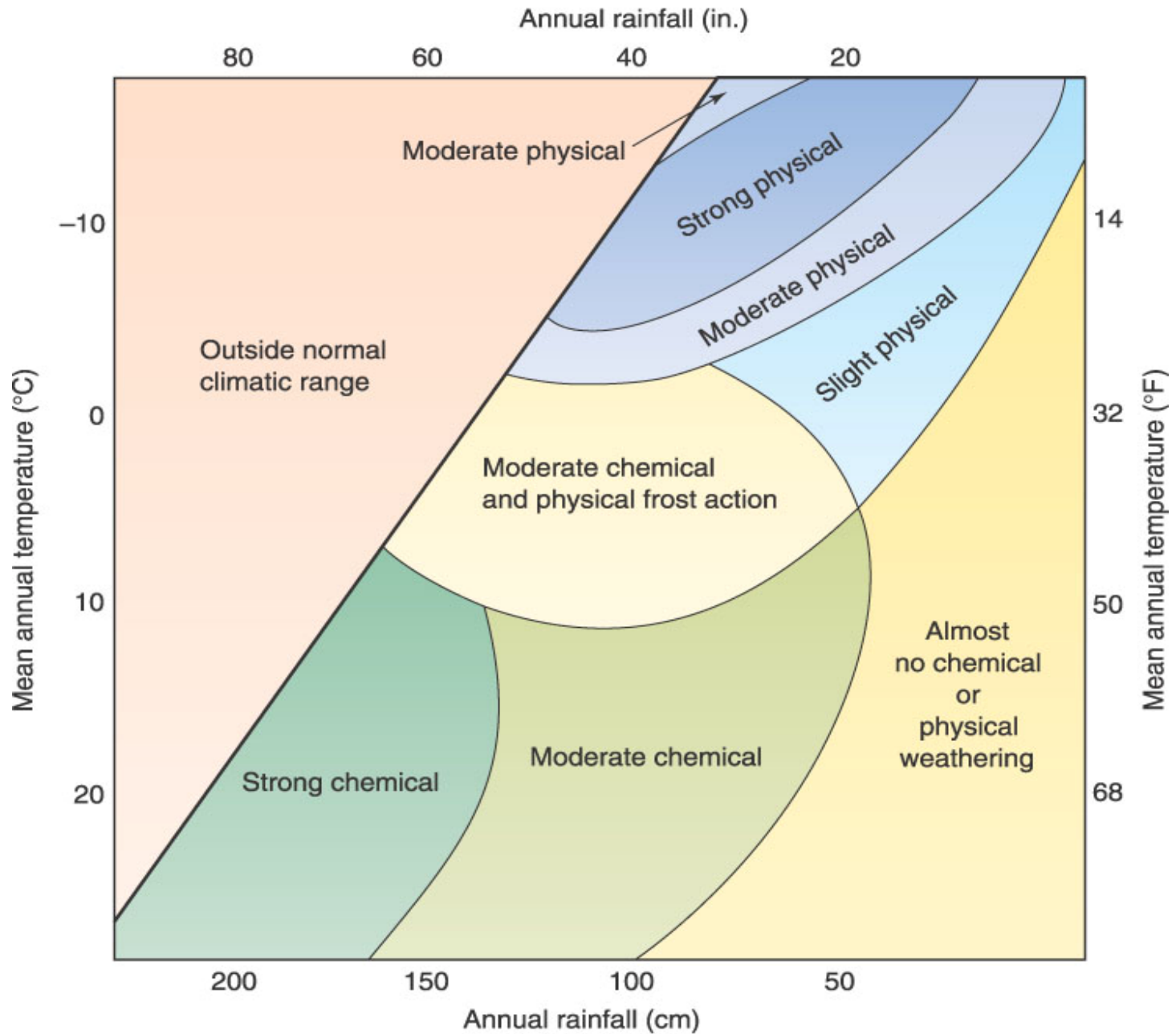
Weathering

- Climate
 - Temperature and moisture characteristics
 - Chemical weathering
 - Most effective in areas of warm, moist climates - decaying vegetation creates acids that enhance weathering
 - Least effective in polar regions (water is locked up as ice) and arid regions (little water)
 - Mechanical weathering
 - Enhanced where there are frequent freeze-thaw cycles

Mechanical and Chemical Weathering

- Fracturing, disintegration caused by mechanical weathering exposes more surface area.
- Greater surface area, means more places for chemical action to occur.

THEORETICAL WEATHERING REGIONS



THEORETICAL WEATHERING REGIONS

